SINTERING BEHAVIOR AND PHYSICAL PROPERTIES OF Bi0.5(Na1-xKx)0.5SnO3 LEAD-FREE CERAMICS

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Abstract. In this study, Bio.5(Na1-xKx)0.5SnO3 (BNKS) ceramics (x = 0, 0.1, 0.2, 0.3, and 0.4) were fabricated via ultrasound wave before milling. The time of ball milling decreased from 20 to 1 h. The X-ray diffraction patterns show that the BNKS has a single-phase structure. When the potassium content increases, the phase structure of the ceramics changes from rhombohedral to tetragonal. When sintered at 1100 °C and x = 0.2, the ceramics' physical properties are the best with the mass density of 5.59 g/cm³, the electromechanical coupling constants k_p of 0,31 and k_t of 0.27, the remanent polarization of 11.9 μ C/cm; the dielectric constant ε_r of 1131, and the highest dielectric constant ε_{max} of 4800.

Keywords: Lead-free ceramics, BNKS, electromechanical coupling factor, sintering, dielectric constant

1 Introduction

The study and application of ferroelectric and piezoelectric ceramics and/or thin films based on PZT have recently been conducted because of the good piezoelectric properties of these materials [1-3]. However, lead-based materials have caused numerous environmental and human health concerns due to the toxicity of lead oxide. Therefore, developing lead-free ceramics and thin film with piezoelectric and ferroelectric properties is of great importance [4-11].

Improving physical properties and reducing sintering temperatures to limit evaporation are two solutions to fabricate leadfree ceramics. The most common way to enhance physical properties is to combine at least two materials with a perovskite ABO₃ structure into a

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multi-component material system the at morphology phase boundary. In addition, for reducing sintering temperatures, selecting the ceramics' proportions to form a liquid phase during sintering is an effective way. In reality, Bio.5(Nao.8Ko.2)0.5TiO3 piezoelectric ceramics with ferroelectric promising and piezoelectric properties have been studied and fabricated [12-17]. However, when Ti4+ is combined with Bi3+, Na⁺, and K⁺, a liquid phase does not form during sintering. In contrast, Sn⁴⁺ can form a liquid phase with Bi3+ at the Eutectic temperature of about 170 °C [18]. Therefore, it is necessary to replace Ti⁴⁺ with Sn4+ or the fabrication of Bio.5(Na1-xKx)0.5SnO3 (BNKS) piezoelectric ceramics with variable x to determine the morphology phase boundary between Bio.5Nao.5SnO3 (BNS) and Bi0.5K0.5SnO3 (BKS).

2 Experimental

Bio.5(Na1-xKx)0.5SnO₃ ceramics, where x = 0, 0.1, 0.2, 0.3, and 0.4, were fabricated from Bi₂O₃ (Merck, 99.5%) Na₂CO₃ (Merck, 99.1%), K₂CO₃ (Merck, 99.1%) and SnO₂ (Merck, 99.6%). The ceramics were produced with the help of ultrasound wave before milling (Fig. 1).

First, Bi₂O₃, Na₂CO₃, K₂CO₃, and SnO₂ were weighed and made smooth with ultrasound (an electric power of 100 W and fixed frequency 28 kHz) in ethanol for one hour. This smoothing technique reduces the time of ball milling from 20 to 1 h [18]. The fine powder was then dried and calcined at 850 °C for 2 h to form BNKS proportions. The BNKS proportions were obtained after the ultrasound and solid-phase reactions according to Eq. (1).

The differential thermal analysis (DTA) and the thermogravimetric analysis (TGA) of BNKS powder, performed at a heating rate of 10 °C/min, show that the phase formation reaction occurs at 823.25 °C (Fig. 2). Therefore, we chose 850 °C as the calcination temperature to ensure that every location in the sample has a temperature of more than 823.25 °C.

The calcined BNKS powder was finely milled for 16 h and pressed in the form of a plate cylinder of a 12 mm diameter and 1.5 mm thickness under a pressure of 100 MPa. The samples were then sintered at 1000, 1050, 1100, and 1150 °C for 2 h.





Fig. 1. Process of fabrication technology of Bio.5(Na1-xKx)0.5SnO3 ceramics



Fig. 2. TG and DTA curves of BNKS powder at 10 °C/min heating rate

The BNKS crystal structure was investigated with X-ray diffraction spectrometry (XRD) at room temperature. The surface morphology of the ceramic samples was determined with field-emission scanning electron microscopy (FESEM; JSM - 6340F), and the density of the ceramics was measured with the Archimedes method. The piezoelectric properties were determined with the resonance method according to the IEEE Standard 61M on an impedance analyzer Agilent 4196B and RLC. The ferroelectric properties were studied with the Sawyer-Tower method.

3 Results and discussion

Fig. 3 shows the mass density of BNKS ceramic samples at different sintering temperatures. When potassium is absent (x = 0), the ceramic has the highest density at 1150 °C (5.14 g/cm³). In contrast, when potassium is incorporated in the ceramic, the material exhibits the highest density (5.59 g/cm³) when sintered at 1100 °C and x = 0.2. This fact proves that the combination of BNS and BKS reduces the sintering temperature, leading to limited evaporation of Bi₂O₃ at high temperatures [18].

Fig. 4 shows the effect of sintering temperatures on the dielectric constant ε and the dielectric loss of tan δ of Bi_{0.5}(Na_{0.8}K_{0.2})_{0.5}SnO₃ ceramics, measured at room temperature at 1 kHz. Accordingly, the dielectric constant increases with the increase in the sintering temperature and reaches the highest value (ε = 1131) at 1100 °C and then decreases gradually. This decrease can be explained via the largest particle size and the density of the ceramics, as shown in Fig. 5. The

larger the particle size is, the larger is the domain with smaller domain boundaries. These changes limit the cracks inside the sample. However, when the temperature is higher than 1100 °C, the particle size remains constant. Still, the evaporation of Bi₂O₃ during sintering at high temperatures would deteriorate the dielectric properties of the ceramics. Therefore, the sintering temperature of 1100 °C is chosen for further experiments.



Fig. 3. Mass density of Bi0.5(Na1-xKx)0.5SnO3 ceramics at different sintering temperatures



Fig. 4. The dependence of the dielectric constant and the dielectric loss on the sintering temperature



Fig. 5. Surface morphology of Bio.5(Nao.8Ko.2)0.5SnO3 ceramics at different sintering temperatures

Fig. 6 shows the X-ray diffraction patterns of BNKS ceramics sintered at 1100 °C. All the samples seem to have a single perovskite structure. The co-existence of rhombohedral and tetragonal phases in the perovskite phase could be analyzed from the XRD patterns with 2 θ from 42 to 47°. Several previous publications demonstrate that the transition between rhombohedral and tetragonal phases could be determined by analyzing the peaks (002)T (tetragonal) and (200)R



Fig. 6. X-ray diffraction patterns of BNKS ceramics

at the proximity of $2\theta \approx 44.6^{\circ}$ [19]. The division of peaks (002)^T and (200)^R indicates the presence of ferroelectric phase transition [20, 21]. Therefore, we believe that both rhombohedral and tetragonal phases exist in the BNKS crystal. When the potassium content increases, the intensity of peaks (002)^T and (200)^T also increases. The (002)^T and (200)^R double peaks are clearly observed when *x* = 0.2. This increase can be further clarified through Goldschmidt's tolerance factors.

$$t_{BNKS} = \frac{r_{Bi} + 0.5((1-x)r_{Na} + xr_K) + r_O}{\sqrt{2}(r_{Sn} + r_O)}$$
(2)

where r is the atomic radius of the element.

When $r_{\text{Sn}} = 0.055$ nm; $r_{\text{Bi}} = 0.103$ nm; $r_{\text{Na}} = 0.102$ nm; $r_{\text{K}} = 0.138$ nm; and $r_0 = 0.14$ nm, the densification factor (t_{BNKS}) is larger than 1. When x increases, the densification factor increases. This increase would cause crystal deformation and may lead to an enhancement in the ferroelectric and piezoelectric properties of the crystals [22].

Fig. 7 (a) shows the temperature dependence of the dielectric constant (ϵ) and the dielectric loss tan δ of the BNKS ceramics sintered at 1100 °C at 1 kHz. This figure shows that this is a relaxer material with the phase transition temperature in the measured temperature range [1]. The dielectric constant increases as xincreases, and when x = 0.2, it reaches the maximum value of 4800. This increase might result from the phase transition near the morphology phase boundary [1]. The first transition is ferroelectric-antiferroelectric phase transition occurring at the depolarization temperature (T_d) , and the second transition is the antiferroelectric-paraelectric phase transition at a higher phase transition temperature (also known as Curie temperature $-T_m$) [20, 23]. Fig. 7 (b) shows the dependence of Td and Tm of BNKS ceramics on the concentration x. As x increases, the Td temperature decreases from 131 to 97 °C while the T_m temperature increases from 245 to 280 °C.

The *P*–*E* hysteresis loops were measured to investigate the ferroelectric properties of BNKS ceramics (Fig. 8a). From these hysteresis loops, the remanent polarization values (*P*_r) and the coercive field (*E*_c) were determined (Fig. 8b). The value of *P*_r reaches a maximum of 11.9 μ C/cm² when *x* = 0.2, and then it decreases gradually. This result is also consistent with the piezoelectric properties investigated in Fig. 9.

To investigate the piezoelectric properties of BNKS ceramics, we recorded the spectra of radial resonance and thick resonance of the samples (Fig. 9). Based on the IRE-61 standard, we calculated the electromechanical coupling factors in the planar mode (k_p) and the thickness mode (k_t).



Fig. 7. (a) Temperature dependence of dielectric constant ε and dielectric loss and (b) dependence on the concentration *x* of two phase-transition temperatures of the BNKS ceramics



Fig. 8. P-E hysteresis loops of BNKS ceramics measured at room temperature



Fig. 9. Spectra of radial resonance (a) and thick resonance (b) of BNKS ceramics

the electromechanical Fig. 10 shows coupling factor ($k_{\rm P}$ and $k_{\rm t}$) and the mechanical quality factor (Q_m) as functions of the concentration x. When x < 0.2, $k_{\rm p}$, $k_{\rm t}$, and $Q_{\rm m}$ increase rapidly with increasing *x*. When x > 0.2, those values start to decrease. Therefore, the piezoelectric properties are optimized when x =0.2; $k_p = 0.31$; $k_t = 0.27$; and $Q_m = 56$. This is also appropriate to the ceramic density and largest particle size. When BNKS is sintered at a low temperature of 1100 °C and x = 0.2, the morphology phase boundary also appears between the tetragonal and rhombohedral phases.



Fig. 10. Electromechanical coupling factors (*k*_P, *k*_t) and mechanical quality factor *Q*^m of BNKS ceramics at room temperature

4 Conclusion

The Bi_{0.5}(Na_{1-x}K_x)_{0.5}SnO₃ ceramics, in which x = 0, 0.1, 0.2, 0.3 and 0.4, were fabricated via the conventional solid-state reaction with the help of ultrasound waves in the milling process. The effects of concentration *x* and sintering temperature on the physical properties of Bio.5(Na1-xKx)0.5SnO3 ceramics were investigated. All the samples seem to have a single perovskite structure. The co-existence of rhombohedral and tetragonal phases in the perovskite phase might improve the dielectric properties of the ceramics. The Bio.5(Nao.8Ko.2)0.5SnO3 ceramics sintered at 1100 °C show the highest dielectric constant ($\varepsilon = 1131$) at room temperature. This is why the BNKS ceramics can have the remanent polarization of 11.9 μ C/cm² and electromechanical coupling factors k_p of 0.31 and k_t of 0.27.

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